

SYNTHESIS OF 2,6-DIMETHYL-6-(8-METHYL-4-METHYLENE-7-NONENYL)-2-CYCLOHEXEN-1-YLMETHANOLS. A COMMENT ON THE STRUCTURE OF MAGYDAR-2,10(20),13-TRIEN-17-OL, THE DITERPENE OF *MAGYDARIS PANACIFOLIA*

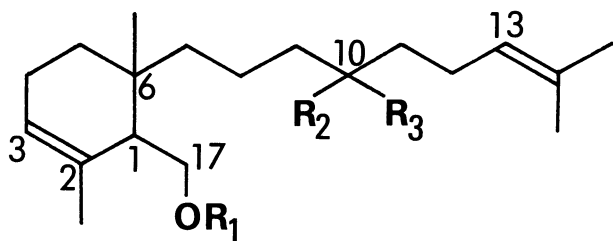
Hajime NAGANO,\* Yukari ISHIKAWA, Yukiko MATSUO, and Michio SHIOTA  
Department of Chemistry, Faculty of Science, Ochanomizu University,  
Otsuka, Bunkyo-ku, Tokyo 112

$\alpha$ -Santonin was stereoselectively converted to a pair of diastereomers of 2,6-dimethyl-6-(8-methyl-4-methylene-7-nonenyl)-2-cyclohexen-1-ylmethanols (Ia and Ib). Both spectral data of Ia and Ib were found to be different from those reported for magydar-2,10(20),13-trien-17-ol isolated from *Magydaris panacifolia* (Vahl) Lange. This indicates that the structure of the natural diterpene must be revised.

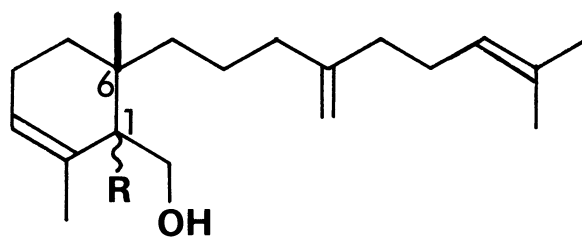
Magydar-2,10(20),13-trien-17-ol (I), its acetate (II), and magydar-2,13-diene-10,17-diol (III) have been isolated from *Magydaris panacifolia* (Vahl) Lange (Umbelliferae) and structurally elucidated by J. de Pascual Teresa *et al.*<sup>1)</sup> However, the stereochemistry of these diterpenes with a novel carbon skeleton has been left undetermined.

In this communication we wish to report a stereoselective transformation of  $\alpha$ -santonin to a pair of diastereomers of 2,6-dimethyl-6-(8-methyl-4-methylene-7-nonenyl)-2-cyclohexen-1-ylmethanols (Ia and Ib), *i.e.*, the two possible stereoisomers for the structure of magydar-2,10(20),13-trien-17-ol (I) or its enantiomer.

Hydroxylation of 5 $\beta$ H-eudesmanolide (1a)<sup>2),3)</sup> ((i) LiN(*i*-Pr)<sub>2</sub>, (ii) MoO<sub>5</sub>.pyridine.HMPPT)<sup>4)</sup> gave a hydroxy lactone (2a) (75% yield; mp 131 - 132.5 °C; 3430 and 1755 cm<sup>-1</sup>;  $\delta$  1.42 (s, C11-CH<sub>3</sub>) and  $\delta$  4.73 (dd, *J* = 10.5 and 4.5 Hz, C6-H)).<sup>5)</sup> Reduction of 2a with LiAlH<sub>4</sub> in THF under reflux afforded a triol (3a) (75% yield; mp 129 - 129.5 °C; 3300 cm<sup>-1</sup>;  $\delta$  3.37 and 3.45 (ABq, *J* = 10.5 Hz, -CH<sub>2</sub>OH)). Oxidative cleavage of 3a with NaIO<sub>4</sub> in aqueous THF gave an oily hydroxy ketone (4a) (98% yield; 3550 and 1710 cm<sup>-1</sup>;  $\delta$  2.18 (s, -COCH<sub>3</sub>)). The Beckmann rearrangement of 4a ((i) *O*-mesitylenesulfonylhydroxylamine, (ii) basic alumina)<sup>6)</sup> gave an amide (5a) (oil; 50%

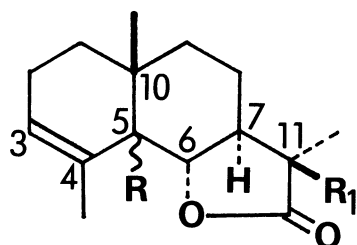


- I     $R_1 = H$              $R_2, R_3 = CH_2$   
II     $R_1 = COCH_3$          $R_2, R_3 = CH_2$   
III     $R_1 = H$                  $R_2 = OH, R_3 = CH_3$

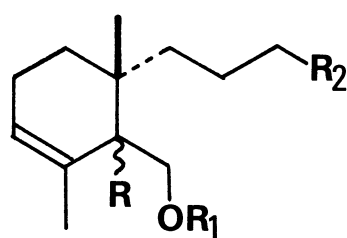


Ia     $R = \beta-H$

Ib     $R = \alpha-H$



- 1a, 1b     $R_1 = H$   
2a, 2b     $R_1 = OH$



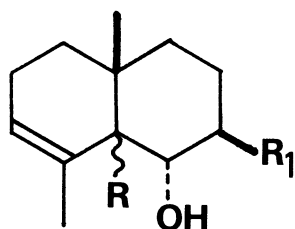
7a, 7b     $R_1 = H$          $R_2 = OH$

8a, 8b     $R_1 = H$          $R_2 = I$

9a, 9b     $R_1 = THP$        $R_2 = I$

11a, 11b     $R_1 = THP$        $R_2 =$

12a, 12b     $R_1 = H$          $R_2 =$



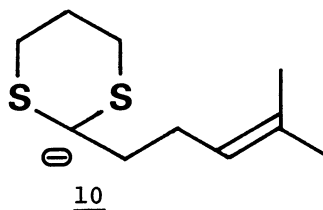
3a, 3b     $R_1 =$

4a, 4b     $R_1 = COCH_3$

5a, 5b     $R_1 = NHCOCH_3$

6a, 6b     $R_1 = NHCH_2CH_3$

13b     $R_1 = THP$        $R_2 =$



1a - 9a, 11a, 12a:  $R = \beta-H$

1b - 9b, 11b - 13b:  $R = \alpha-H$

yield; 1645 and 1545  $\text{cm}^{-1}$ ;  $\delta$  2.02 (s,  $-\text{NHCOCH}_3$ ), which was then reduced with  $\text{LiAlH}_4$  to afford a hydroxy amine (6a) (oil; 80% yield; 3400  $\text{cm}^{-1}$ ;  $\delta$  1.12 (t,  $J = 7$  Hz,  $-\text{NHCH}_2\text{CH}_3$ ) and  $\delta$  2.69 (q,  $J = 7$  Hz,  $-\text{NHCH}_2\text{CH}_3$ );  $m/z$  223,  $\text{M}^+$ ). Oxidative cleavage of 6a with  $\text{NaIO}_4$  in aqueous THF gave an unstable dialdehyde, which was reduced, without isolation, with  $\text{NaBH}_4$  at 0 °C to an oily diol (7a) (3350  $\text{cm}^{-1}$ ;  $\delta$  2.27 (br. s, OH) and  $\delta$  3.65 (m,  $-\text{CH}_2\text{OH}$ )).

Completion of the side chain transformation was brought as follows. The diol (7a) was partially tosylated with 1 molar equivalent of *p*-TsCl in pyridine at 5 °C. The resulting reaction mixture was then treated with NaI in acetone under reflux to give a monoiodide (8a) (oil; 18% overall yield from 6a;  $\delta$  0.86 (s,  $\text{C}_{10}\text{-CH}_3$ ),  $\delta$  3.19 (t,  $J = 6.5$  Hz,  $-\text{CH}_2\text{-I}$ ), and  $\delta$  3.72 (d,  $J = 3.5$  Hz,  $-\text{CH}_2\text{OH}$ )<sup>7)</sup> as the major product. Tetrahydropyranylation of 8a with dihydropyran in the presence of *p*-TsOH gave 9a in quantitative yield. Treatment of 9a with the carbanion (10) generated from 2-(4-methyl-3-pentenyl)-1,3-dithiane<sup>8)</sup> with *n*-butyllithium gave 11a (oil; 52% yield),<sup>9)</sup> which on hydrolysis with copper(II) chloride-copper(II) oxide in acetone<sup>10)</sup> afforded an oily hydroxy ketone (12a) (40% yield; 3470 and 1710  $\text{cm}^{-1}$ ;  $m/z$  274,  $[\text{M} - \text{H}_2\text{O}]^+$ ). Finally, 12a was methylenated with  $\text{Ph}_3\text{PCH}_2$  to give Ia (oil; ca. 30% yield; 3450 and 885  $\text{cm}^{-1}$ ;  $m/z$  290,  $\text{M}^+$ ),<sup>11)</sup> whose  $^1\text{H}$  NMR spectral data were obviously different from those reported for I (Table).

Table  $^1\text{H}$  NMR spectral data ( $\text{CDCl}_3$ ,  $\delta$  values)

	$\text{C6-CH}_3$	$\text{C=C-CH}_3$	$\text{CH}_2\text{OH}$	$\text{C=CH}_2$	$\text{CH=C(CH}_3)_2$	$\text{C3-H}$
<u>I</u>	0.71 (s)	1.53 (s, 6H) 1.63 (s, 3H)	3.65 (ABdq)	4.76 (s) 4.62 (s)	5.03 (m)	5.03 (m)
<u>Ia</u>	0.85 (s)	1.62 (s) 1.70 (s) 1.74 (s)	3.73 (m)	4.72 (s)	5.16 (m)	5.62 (m)
<u>Ib</u>	0.99 (s)	1.60 (s) 1.69 (s) 1.74 (d, $J =$ 1.5 Hz)	3.73 (m)	4.70 (s)	5.11 (m)	5.61 (m)

5 $\alpha$ H-Eudesmanolide (1b)<sup>3)</sup> was converted to an alcohol (8b)<sup>7)</sup> (12% overall yield;  $\delta$  1.00 (s,  $\text{C}_{10}\text{-CH}_3$ ),  $\delta$  3.13 (t,  $J = 7$  Hz,  $-\text{CH}_2\text{-I}$ ), and  $\delta$  3.72 (d,  $J = 3$  Hz,  $-\text{CH}_2\text{OH}$ ) by the same procedure described above.<sup>12)</sup> 8b was then transformed to a tetrahydropyranyl ether (9b), which was then treated with 2-lithio-1,3-dithiane at 0 °C to give 13b (oil; 34% yield from 8b). The carbanion generated from 13b with *n*-butyllithium was treated with 4-methyl-3-pentenyl iodide<sup>8)</sup> to give 11b (oil; 60% yield).<sup>9)</sup> Hydrolysis of 11b with copper(II) chloride-copper(II) oxide in acetone<sup>10)</sup> gave a hydroxy ketone (12b) (oil; 40% yield; 3450 and 1710  $\text{cm}^{-1}$ ). Methylenation

of 12b with  $\text{Ph}_3\text{PCH}_2$  afforded Ib (oil; ca. 25% yield; 3400 and 890  $\text{cm}^{-1}$ ;  $m/z$  290,  $\text{M}^+$ ).<sup>11)</sup> The  $^1\text{H}$  NMR spectral data of Ib were clearly different from those reported for I (Table).

We therefore suggest that the structure of the natural diterpenes (I, II, and III) should be revised.

We wish to thank Professor Akio Kobayashi, Laboratory of Food Chemistry, Ochanomizu University, for the mass spectral measurements, and Dr. Hiroshi Hirota, Faculty of Science, The University of Tokyo, for the FT-NMR measurements.

#### References

- 1) J. de Pascual Teresa, C. Grande, and M. Grande, *Tetrahedron Lett.*, **1978**, 4563.
- 2) Numbering of the eudesmane skeleton is used for all the compounds derived from eudesmanolides (1a and 1b) except for Ia and Ib.
- 3) 5 $\beta$ H-Eudesmanolide (1a) was prepared from  $\alpha$ -santonin following the procedure described by A. E. Greene. A. E. Greene, *J. Am. Chem. Soc.*, **102**, 5337 (1980). 5 $\alpha$ H-Eudesmanolide (1b) was also prepared from  $\alpha$ -santonin using a slightly modified literature procedure. W. Cocker and T. B. H. McMurry, *J. Chem. Soc.*, **1956**, 4549.
- 4) E. Vedejs, *J. Am. Chem. Soc.*, **96**, 5944 (1974); E. Vedejs, D. A. Engler, and J. E. Teleschow, *J. Org. Chem.*, **43**, 188 (1978).
- 5) The configuration of the hydroxyl group at C-11 was assigned to be  $\beta$  on the basis of the precedents similar to this case. S. Yamada, K. Nakayama, and H. Takayama, *Tetrahedron Lett.*, **1981**, 2591 and references cited therein.
- 6) Y. Tamura, H. Fujiwara, K. Sumoto, M. Ikeda, and Y. Kita, *Synthesis*, **1973**, 215.
- 7) The  $^1\text{H}$  NMR spectral data of 8a and 8b are consistent with those of  $\alpha$ -cyclogeraniol. M. Shibasaki, S. Terashima, and S.-I. Yamada, *Chem. Pharm. Bull.*, **23**, 272 (1975).
- 8) W. Biernacki and A. Gdula, *Synthesis*, **1979**, 37.
- 9) E. J. Corey and D. Seebach, *Angew. Chem. Int. Ed. Engl.*, **4**, 1075 (1965).
- 10) K. Narasaka, T. Sakashita, and T. Mukaiyama, *Bull. Chem. Soc. Jpn.*, **45**, 3724 (1972).
- 11) Ia and Ib were isolated by preparative TLC ( $\text{SiO}_2$ , hexane-ether (2:1),  $R_f(\text{Ia}) = 0.66$  and  $R_f(\text{Ib}) = 0.5$ ). GLC (20% Carbowax 20M, 3 mm $\times$ 2 m, 195  $^\circ\text{C}$ ,  $\text{N}_2$  (1.2 Kg/cm $^2$ )):  $t_R(\text{Ia}) = 9.0$  min and  $t_R(\text{Ib}) = 9.2$  min. Both Ia and Ib were contaminated with a more volatile component (<10%,  $t_R = 4.5$  min). MS: Ia,  $m/z$  290(7%,  $\text{M}^+$ ), 272(8,  $[\text{M}-\text{H}_2\text{O}]^+$ ), 135(23), 123(20), 122(23), 121(38), 110(24), 109(63), 107(54), 95(37), 93(46), 81(46), 79(20), and 69(100); Ib,  $m/z$  290(5,  $\text{M}^+$ ), 272(12,  $[\text{M}-\text{H}_2\text{O}]^+$ ), 149(26), 135(27), 122(52), 121(48), 119(27), 109(46), 107(94), 95(48), 93(54), 91(23), 81(57), 79(23), and 69(100).
- 12) Spectral data of 2b - 7b are shown below. 2b,<sup>5)</sup> mp 185 - 186.5  $^\circ\text{C}$ , 3470 and 1755  $\text{cm}^{-1}$ ,  $\delta$  1.44 (s, C11- $\text{CH}_3$ ) and  $\delta$  4.37 (dd,  $J = 12$  and 9 Hz, C6-H); 3b, oil, 3350  $\text{cm}^{-1}$ ,  $\delta$  3.37 and 3.53 (ABq,  $J = 12$  Hz,  $-\text{CH}_2\text{OH}$ ); 4b, mp 66 - 67  $^\circ\text{C}$ , 3420 and 1708  $\text{cm}^{-1}$ ,  $\delta$  2.18 (s,  $-\text{COCH}_3$ ); 5b, mp 125 - 127  $^\circ\text{C}$ , 1655 and 1545  $\text{cm}^{-1}$ ,  $\delta$  2.01 (s,  $-\text{NHCOCH}_3$ ),  $m/z$  237,  $\text{M}^+$ ; 6b, mp 64 - 65  $^\circ\text{C}$ , 3400  $\text{cm}^{-1}$ ,  $\delta$  1.07 (t,  $J = 7$  Hz,  $-\text{NHCH}_2\text{CH}_3$ ) and  $\delta$  2.3 - 2.9 (m, C7-H and  $-\text{NHCH}_2\text{CH}_3$ ),  $m/z$  223,  $\text{M}^+$ ; 7b, oil, 3340  $\text{cm}^{-1}$ ,  $\delta$  3.5 - 3.8 (m,  $-\text{CH}_2\text{OH}$ ).